

## Chapter 4 Atomic Structure Wordwise Answers

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Dalton's Atomic Theory (experiment based!) 3) Atoms of different elements combine in simple whole-number ratios to form chemical compounds 4) In chemical reactions, atoms are combined, separated, or rearranged – but never changed into atoms of another element. 1) All elements are composed of tiny indivisible particles called atoms

### Chapter 4 Atomic Structure

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Name Chapter 4: Atomic Structure Worksheet Answer the ...  
Start studying Science Wordwise chapter 4. Learn vocabulary, terms, and more with flashcards, games, and other study tools. Search. Create. Log in Sign up. Log in Sign up. Science Wordwise chapter 4. ... atomic number. sum of the protons and neutrons in the nucleus of an atom. mass number. atoms of the same element having different numbers of ...

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atomic number: equals the number of protons in an atom of that element: mass number: the sum of the protons and neutrons in the nucleus of that atom: isotopes: atoms of the same element that have different numbers of neutrons and different mass numbers: energy levels: the possible energies that electrons in an atom can have: electron cloud

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Chapter 4 Atomic Structure WordWise Solve the clues to determine which vocabulary terms from Chapter 4 are hidden in the puzzle. Then find and circle the terms in the puzzle. The terms may occur vertically, horizontally, or diagonally. em a s s n umb e run nl o rb i t a l x api r e e n l t p t s p b k s gn a c a s r d c r hl o be l dt g o f l s gat

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Section 4.1 Studying Atoms Flashcards | Quizlet

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a vertical column of elements in the periodic table; the constituent elements of a group have similar chemical and physical properties.

Unit 2 - Chapters 4, 5 & 6 - Mrs. Gingras' Chemistry Page

The atomic mass of an isotope is the sum of the number of protons and neutrons in its nucleus. The max. number of electrons in the first energy level of an atom is:2. in the second energy level is:8. in the third energy level is:18. A certain atom has 26 protons, 26 electrons, and 30 neutrons.

Chemistry- Ch. 4 Vocab- Atomic Structure Flashcards | Quizlet

Chapter 4: Atomic Structure Worksheet . Answer the following questions, circle the best answer. 1) Which of the following are isotopes of each other? a)  $^{14}\text{C}$  and  $^{14}\text{N}$  b)  $^3\text{H}$  and  $^4\text{He}$  c)  $^2\text{H}$  and  $^1\text{H}$  d) none of these . 2) The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . a)  $5+$  b)  $-3$  c)  $3+$  d)  $-9$

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Chapter 4: The Structure of the Atom

Chapters 4, 5 & 6. This unit's focus is on atomic structure. Students will cover concepts related to atomic mass, isotopes, electron configurations and periodicity.

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Chapter 4 Atomic Structure - Henry County School District

104 Chapter 4 • The Structure of the Atom Figure 4.2 In his book A New System of Chemical Philosophy, John Dalton presented his symbols for the elements known at that time and their possible combinations.

Guided Reading And Study Workbook Chapter 4 Atomic Structure

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