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Answer questions 1-6 on page 802; Complete the interpreting the graph sections on pages 803 & 804. 04/22/2015 Describe what happens during a nuclear chain reaction. Explain neutron moderation, and neutron absorption. Explain what nuclear waste is, how it's created and what is done with it after it is replaced.

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Chapter 25 Section 25.2 (continued) Half-Life Discuss

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Explain that, for each element, there exists only a small range of neutron-to-proton ratios that produce stable nuclei. If a nucleus does not reflect a stable ratio, it spontaneously decays until a stable ratio of neutrons to protons results. Relate Explain that the nuclear stability that

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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down.

•The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive

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isotope is an unstable atom which breaks down on its own, releasing energy and/or

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Quia - Chapter 25 "Nuclear Chemistry"

Nuclear Chemistry Nuclear Transformations •

Rutherford in 1919 performed the first nuclear transformation. • The transmutations are sometimes

represented by listing in order, the target nucleus, the

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bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes: ${}_{17}^{14}N + {}_2^4He \rightarrow {}_{17}^{18}O + {}_1^1H$

Chapter 25: Nuclear Chemistry - Ms. Kots' Chemistry Page

The production of energy in a nuclear reactor can be stopped by pulling out all control rods. A breeder reactor produces more fuel than it uses. The fission products produced in nuclear power plants are not radioactive. An uncontrolled chain reaction led to the nuclear accident in Chernobyl, Ukraine. Chemistry: Matter and Change Chapter 25 149

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25.2 Nuclear Transformations 25

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A B; beta particle: a fast-moving electron formed by the decomposition of a neutron: half-life: the time required for one-half of the atoms of a radioisotope to emit radiation and to decay to products

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