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Chapter 1—The Atomic Nature of Matter: Selected Questions ...

TEST - THE ATOMIC NATURE OF MATTER - MONDAY, FEBRUARY 12, 2007 2/2/07 OBJECTIVE: The Atomic Nature of Matter HOMEWORK: Study for quiz on the atomic nature of matter; p. 256 1-10 Chapter 17 Review Questions TEST - THE ATOMIC NATURE OF MATTER - MONDAY, FEBRUARY 12, 2007

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## Exercise Answers

2/5/07 OBJECTIVE: Quiz - The Atomic Nature of Matter; Drawing Atoms

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MATTER

Chapter 17--The Atomic Nature of Matter. a

\_\_\_\_\_ is a substance that is made of atoms of different elements combined in a fixed proportion.

Chapter 17 The Atomic Nature  
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Exercises - MoHS CORE 1 Program

3 Broughton High School of Wake County Chapter 17 Physical Science 2017 Mr. Davis Properties of Atoms and the Periodic Table

Section 1: Structure of the Atom – (pages 506 – 511) 1. \_\_\_\_\_ are the abbreviated in scientific shorthand – first letter of two of elements' name.

Chapter 17 The Atomic Nature of Matter Flashcards | Quizlet

CHAPTER 17 THE ATOMIC NATURE OF MATTER 325. 17.1 Elements. Just as dots of light of only three colors

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combine to form almost every conceivable color on a television screen, only about 100 distinct kinds of atoms combine to form all the materials we know about. Atoms are the building blocks of matter.

## Chapter 17

Chapter 17: The Atomic Nature of Matter Questions.

Description. Chapter 17: The Atomic Nature of Matter Questions. Total Cards. 21.

Subject. Physics. Level.

11th Grade. Created.

12/31/2011. ... Atomic

number of an element is determined by the number of \_\_\_\_\_ in the nucleus.

Definition. protons:

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Physics Powerpoints - Mr. Jeremy T. Rosen  
Conceptual Physics Flipped Classroom, The Atomic Nature

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## Exercise Answers

Conceptual Physics (17.1 -17.4) Building blocks of matter A material composed of only one kind of atom The perpetual jiggling of particles that are just large enough....

Concept-Development 17-1

Practice Page

chapter 11: rotational

mechanics chapter 12:

universal gravitation

chapter 13: gravitational

interactions chapter 14:

satellite moon chapter 15:

special relativity- space &

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relativity-length, momentum

and energy chapter 17: the

atomic nature of matter

chapter 18: solids chapter

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19. liquids chapter 20:  
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Summary - MoHS CORE 1  
Program

The Atomic Nature of Matter  
Use the periodic table on  
page 336 of your text to  
help you answer the  
following questions. 1. When

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the atomic nuclei of hydrogen and lithium are squashed together (nuclear fusion) the element that is produced is 2. When the atomic nuclei of a pair of lithium nuclei are fused, the element produced is 3.

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17.7 The Atomic Nucleus (pages 331–333) 31. Where is most of the mass of an atom located? 32. Describe how Ernest Rutherford discovered that an atom is composed mostly of empty space. Match each phrase with the correct word or words. 33. principal building blocks of an atom's nucleus 34. nucleons in an

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electrically neutral state

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Term atom Definition The smallest particle of an element that can be identified with that element. Consists of protons and neutrons in a nucleus surrounded by electrons.

Conceptual Physics Chapter 17 Part 1

Chapter 1—The Atomic Nature of Matter: Selected Questions for Practice Exam.  
6. Substances that cannot be decomposed into two or more simpler substances by chemical means are called a.

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pure substances.

b.compounds. c. molecules.

d.elements. e. none of these

7.A pure solid decomposes on heating, yielding a solid and a gas, each of which is a ...

Chapter 17: The Atomic Nature of Matter Questions Flashcards

Conceptual Physics Reading and Study Workbook N Chapter 17 135 17.8 Electrons in the Atom The arrangement of electrons in the shells about the atomic nucleus dictates the atom's chemical properties. • Electrons that orbit the atomic nucleus are identical to the electrons that flow in the wires of

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electric circuits.

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