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Work and Heat Heat

is the transfer of

thermal energy from

one object to another

because of a

difference in

temperature.

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Heat. the transfer of thermal energy from one object to another as the result of a difference in temperature. True.

T/F: On the Celsius Scale, the reference points for temperature are the freezing and boiling points of water. thermal energy

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most scientists  
thought that heat was

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a fluid called caloric  
that flowed between  
objects.

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concluded that heat is  
not a form of matter.

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## Energy And Heat 474 Chapter 16 FOCUS Objectives

16.1.1 Explain how heat and work transfer energy.

16.1.2 Relate thermal energy to the motion of particles that make up a material.

16.1.3 Relate temperature to thermal energy and to thermal expansion.

16.1.4 Calculate thermal energy,

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temperature change,  
or mass

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Grygla Public School  
the study of  
conversions between  
thermal energy and  
other forms of energy:  
heat engine: any  
device that converts  
thermal energy into  
work: waste heat:  
thermal energy  
discharged into an  
area at a lower

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temperature without  
being converted into  
useful work: external  
combustion engine: a  
heat engine in which  
the fuel burns outside  
the engine

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Section 16.2 Heat and

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Thermodynamics  
(pages 479–483) This  
section discusses  
three kinds of thermal  
energy transfer and  
introduces the first,  
second, and third  
laws of  
thermodynamics.

Reading Strategy  
(page 479) Build  
Vocabulary As you  
read this section, add  
definitions and

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examples to complete  
the table.

Chapter 16 - Chapter  
16 Thermal Energy  
and Heat 16.1 ...

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Energy and Heat.  
16.1 Thermal Energy  
and Matter Heat  
transfer of thermal  
energy from one  
object to another  
because of temp.

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difference Heat flows spontaneously from hot objects to cold objects Temperature measure of how hot or cold an object is compared to a ref. pt. chapter 16 thermal energy and heat | Heat (356 views)

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Heat

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Section 16.1 Thermal  
Energy and Matter  
(pages 474–478) This  
section defines heat  
and describes how  
work, temperature,  
and thermal energy  
are related to heat.  
Thermal expansion  
and contraction of  
materials is  
discussed, and uses

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of a calorimeter are  
explained. Reading  
Strategy (page 474)

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Harris Science

16.1 Thermal Energy  
and Matter Heat flows  
spontaneously from  
hot objects to cold  
objects. □ Heat is the  
transfer of thermal  
energy from one

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object to another  
because of a  
temperature  
difference.

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your knowledge of  
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## Chapter 16 - Thermal Energy

The quantity of energy transferred by heat from a body depends on a number of physical properties of the body and its surroundings. For a given substance, the rate at which thermal energy is transferred

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by conduction depends on temperature difference, cross-sectional area, and a thermal conductivity constant that is unique to the substance.

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Christopher 20  
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Chapter 15

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Jaspreet 20 Terms.

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Terms. meanup. Ch.  
13 - Forces in fluids  
11 Terms.

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Work and Heat  
Temperature is the  
measure of how hot  
or cold something is  
compared to a  
reference point. The

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Celsius scale has reference points of freezing and boiling points of water On the Kelvin scale the reference point is absolute zero

Absolute Zero is the temperature at which molecules essentially stop (no kinetic energy)

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