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Chapter 16
Answer Key
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Chapter 4, Lesson 4:
Energy Levels,
Electrons, and ...
Look over the Study
Guide at the end of the
chapter. Covalent
Bonding Chapter Wrap-
Up Explain how
covalent bonds in

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carbon account for the vast number of carbon compounds, including those responsible for living organisms. Accept all reasonable responses. Answers should indicate that carbon, like all elements in its

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Answer Key

COVALENT BONDING

Class 8.2 8.2 8.2 8.2 8.3

8.3 8.3 8.3 195 Chapter

Quiz Choose the best

answer and write its

letter on the line. 1. A

bond in which each

atom contributes two

electrons is a. a double

covalent bond. b. an

ionic bond. c. a polar

covalent bond. d. a

coordinate covalent

bond. 2. The electron

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Covalent Bonding

dot structure for
hydrogen sulfide, H_2S ...

chemistry chapter 16

covalent bonding

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Chapter 8 Concepts of
Chemical Bonding .

Chemical Bonds Three

types: – Ionic

Electrostatic attraction

... • Lewis structures for

16 elements • It is rare

to use Lewis pictures for

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Answer Key

other ... bonding and nonbonding. Lines correspond to 2 electrons in bond .

Covalent Bonding • A bond where electrons from each atom are shared • Each ...

Covalent Bonding

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6 Chemical Bonding

Chapter 8 Covalent

Bonding and Molecular

Structure 8-4 H 2

molecule. More

sophisticated

descriptions of chemical

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bonding will be discussed in Chapter 9.

8.3 Lewis Structures

OWL Opening

Exploration 8.X One of the most important tools chemists use to predict the properties of a chemical species is its Lewis structure.

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Chapter 16

Covalent Bonding

Chapter 16: Covalent
Bonding. Chemistry

vocab and whatnot.

STUDY. PLAY. ... a

covalent bond in which

the bonding electrons

are shared equally by

the bonded atoms,

resulting in a balanced

distribution of electrical

charge ... Chapter 15:

Ionic Bonding and Ionic

Compounds. 80 terms.

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Chapter 13: Electrons
in Atoms. 60 terms.

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35. Explain why compounds containing C-N and C-O single bonds can form coordinate covalent bonds with H^+ but compounds containing only C-H and C-C bonds cannot. 16.1 An unshared pair of

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Answer Key

electrons is needed for a coordinate covalent bond. There are no unshared pairs in C-H and C-C bonds. 36.

Chapter 16 Covalent
Bonding Answer

Chapter 16 – Covalent
Bonding Chapter 16: 1
– 26; 28, 30, 31, 35-37,
40, 43-46, Extra Credit:
50-53, 55, 56, 58, 59,

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Covalent Bonding

Answer Key

62-67 Section 16.1 –

The Nature of Covalent
Bonding Practice

Problems 1. Draw

electron dot structures

for each molecule. a.

chlorine b. bromine c.

iodine What do you

observe about the three

structures? 2.

Chapter 16 Review

Answers -

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Chapter 16

Covalent Bonding

Answer key 8 covalent
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results. guided reading
and study

workbook/chapter 2 13

naming covalent

compounds answer key

owners. atoms and

bonding guided reading

and study covalent

bonds this section

describes how chemical

bonds 16 COVALENT

BONDING guided

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Covalent Bonding
Answer Key

Section Vocabulary -
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a(n) bond. 5. A covalent bond in which two atoms share three pairs of electrons is called a(n) bond. 6. A bond in which two atoms share one pair of electrons is a bond. 7. A nonbonding pair of electrons in the valence shell of an atom is called a(n) pair.

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Answer the following items in the space provided. 8.

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Chapter 8: Covalent
Bonding and Molecular
Structure

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319 Chapter 4, Lesson
4: Energy Levels,

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Electrons, and Covalent
Bonding Key Concepts •

The electrons on the outermost energy level of the atom are called valence electrons.

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Chapter 16 Review
Answers ... Atoms of
most elements quickly
form chemical bonds

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Answer Key

with other . atoms to make more stable compounds. 3. A covalent bond holds a water molecule together. 4. A molecule. 5. The subscript tells you the ratio of atoms of each element in a compound. In .

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that are introduced in this section. Each blank can be completed with a term, short phrase, or number The quantum mechanical model of bonding assumes that 1.

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flC) r rh atomic orbitals overlap to produce 1 A molecular orbit that 2-can be occupied by two electrons of a covalent bond is called a 3. S

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bonding work, Chemical
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Bing

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and Change Name Date

Class Name Date Class

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CONTENT MASTERY

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Covalent Bonding

Section 9.2 Naming

Molecules In your

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how binary compounds and acids are named from their formulas.

Chapter 8 Concepts of Chemical Bonding

CHAPTER 6 REVIEW

Chemical Bonding

SECTION 2 SHORT

ANSWER Answer the

following questions in the space provided. 1.

Use the concept of potential energy to

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describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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