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Joseph_Wang6. Terms in this set

(233) Water is an amazing liquid that makes. plant and animal life on Earth Possible. In addition to water forming the Earth's oceans water, it

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also forms the.

Quia - Chapter 15 "Water and Aqueous Systems"
Chapter 15 (Water and Aqueous Systems) Test Study Guide ? The bonds between the hydrogen and oxygen atoms in a water molecule

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are polar covalent bonds. ? The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. ? The bonds between adjacent water molecules are called hydrogen bonds.

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the possible dependence of percent
ionization of acetic acid with

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concentration. We are to choose the one that best represents the trend for this weak acid.

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Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its

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Properties - 15.1 Lesson Check -
Page 493: 1 Answer The hydrogen bonding in water cause the high surface tension and low vapor pressure of water.

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Chapter 15 - Water and Aqueous
Systems - 15.3 Heterogeneous
Aqueous Systems - 15.3 Lesson
Check - Page 507: 21 Answer
Colloids cannot be separated
through filtering because their
particles are smaller than those of

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suspensions and they do not settle out over time.

Chemistry (12th Edition) Chapter 15
- Water and Aqueous ...

The high surface tension of water is due to the: a. small size of water molecules. b. low mass of water

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molecules. c. hydrogen bonding between water molecules. d. covalent bonds in water molecules.

_____ 12. Salts and other compounds that remove moisture from air are said to be: a. efflorescent. c. colloidal. b. surfactant. d. hygroscopic. 10 9 ...

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SECTION 15.1 WATER AND ITS

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PROPERTIES (pages 445–449)

1. Chapter 15 “Water and Aqueous Systems” Pre-AP Chemistry

Charles Page High School Stephen

L. Cotton 2. Section 15.1 Water and
it’s Properties OBJECTIVES:

–Explain the high surface tension
and low vapor pressure of water in

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terms of the structure of the water molecule and hydrogen bonding. 3.

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helps students learn the essential lessons associated with water and aqueous systems.

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Chapter 15 Water and Aqueous
Systems 159 SECTION 15.1 WATER

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AND ITS PROPERTIES (pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. Water in the Liquid State (pages 445–447) 1.

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CHAPTER 15 | Aqueous Equilibria:
Chemistry of the Water World
The Water Molecule: a Review
Water is a simple tri-atomic
molecule, H_2O . Each O-H bond is
highly polar, because of the high
electronegativity of the oxygen (N,
O, F, and Cl have high values) bond

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angle of water = 105° due to the bent shape, the O-H bond polarities do not cancel. This means: water is a polar molecule.

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AND AQUEOUS SYSTEMS & nbs p;
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Quizzes. What Animal Would Your

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Chapter 13 - States of Matter

Chapter 14 - Behavior of Gases

Chapter 15 - Water and Aqueous
Systems Chapter 16 - Solutions

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Chapter 17 -Thermochemistry

Chapter 18 - Reaction Rates and

Equilibrium Chapter 19 - Acids,

Bases and Salts Chapter 20 -

Oxidation-Reduction Reactions

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