

## Chapter 14 Review Acids And Bases Mixed

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Chemistry Test Chapter 14 Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. ashleysanders47. Terms in this set (51) Acids. reacts with base to form salt and water. Starts with h+ (hydrogen ion). Binary Acid. an acid that contains only two different elements: hydrogen and one of the more electronegative ...

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CHAPTER 14 REVIEW . Acids and Bases. SHORT ANSWER Answer the following questions in the space provided.

1. a. Write the two equations that show the two-stage ionization of sulfurous acid in water. stage 1:

$\text{H}_2\text{SO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{HSO}_3^-(\text{aq})$  b. Which stage of ionization usually produces more ions?

Explain your answer.

### Chapter 14 - Acids and Bases Flashcards

Chemistry: The Molecular Science (5th Edition) answers to Chapter 14 - Acids and Bases - Conceptual Exercise 14.1 - Bronsted-Lowry Acids and Bases - Page 609 b including work step by step written by community members like you. Textbook Authors: Moore, John W.; Stanitski, Conrad L., ISBN-10: 1285199049, ISBN-13: 978-1-28519-904-7, Publisher: Cengage Learning

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CHAPTER 14 REVIEW Acids and Bases - Weebly CHAPTER FOURTEEN ACIDS AND BASES CHAPTER 14 ACIDS AND BASES 5 When  $\text{H}_3\text{PO}_4$  is added to water, the three acids that are present are  $\text{H}_3\text{PO}_4$ ,  $\text{H}_2\text{PO}_4^-$ , and  $\text{HPO}_4^{2-}$ .  $\text{H}_3\text{PO}_4$ , with the largest  $K_a$  value, is the strongest of these weak acids The conjugate [DOC] Chapter 14 Review Acids And Bases Mixed CHAPTER 14 ...

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Interactive Example 14.1 - Acid Dissociation (Ionization) Reactions Write the simple dissociation (ionization) reaction (omitting water) for each of the following acids: a. Hydrochloric acid ( $\text{HCl}$ ) b. Acetic acid ( $\text{HC}_2\text{H}_3\text{O}_2$ ) c. The anilinium ion ( $\text{C}_6\text{H}_5\text{NH}_3^+$ ) d. The hydrated aluminum(III) ion [ $\text{Al}(\text{H}_2\text{O})_6$ ] $^{3+}$

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Chapter 14 - Acids and Bases. Description. Chapter 14 Review. Total Cards. 36. Subject. Chemistry. Level. 11th Grade. Created. 04/25/2013. ... General Properties of Acids: Definition. 1) sour taste, corrosive 2) characteristically changes acid-base indicator 3) react with active metals to form hydrogen gas 4) react with bases to form salt and water

### Chemistry Test Chapter 14 Review Flashcards | Quizlet

Acids and Bases: Chapter 14 & 15 . HW: •Read Ch 14: •Fill in as much of the acid base table as you can, as you read . Acid base conductivity and reactivity Conductivity, Reac'tivity,, ... 3.Work on ch 14-15 review (due Wednesday, test Friday!) • Solutions in which [H

### Chapter 14 Review Acids And Bases Mixed

Arrhenius Acid -a hydrogen containing compound that ionizes to produce hydrogen ions ( H<sup>+</sup>) when dissolved in water Because molecular acids are not made of ions, they cannot dissociate. HCl (aq ) → H<sup>+</sup>(aq ) + Cl<sup>-</sup>(aq ) They must be pulled apart, or ionized, by the water. Chapter 14: Acids and Bases Ch 14 Page 1

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### Chapter Fourteen [Acids and Bases] - Wattsburg

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