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water, whole wheat bread, or sugar. whole wheat bread. all of the following are homogeneous mixtures except: soil, sugar water, gasoline, or salt water.

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Molarity. a unit describing CONCENTRATION of a solution.

solutions. homogeneous mixture of 2 or more pure

substances. a solution is composed of. solvent & solute.

solvent. in greater quantity than solute, typically a liquid but not always. solute.

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reactants. On the left side of a chemical equation, you have the?

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Chapter 12 - Chemical Kinetics . 12.1 Reaction Rates . A.
Chemical kinetics 1. Study of the speed with which reactants are converted to products B. Reaction Rate 1. The change in concentration of a reactant or product per unit of time [] t A t concentration of A at time t concentration of A at time t Rate ' ? = ? ? = 2 1 2 1. a ...

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Chemistry Chapter 12 Study Guide Flashcards by ProProfs 174 Study Guide for An Introduction to Chemistry a model (and therefore a simplification of reality), it is extremely useful. You will find the information in Table 12.1: Covalent Bonding Patterns very helpful when drawing Lewis structures (the task described in Section 12.2).

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