

### Chapter 11 Review Gases Answer Key

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Chapter 11 Review Gases Section 4 Answers  
Chapter 11 Review Gases Section 1 Answer Key Chapter 11 Review Gases Section Chapter 11 Review Gases Section (Section 10.2) Review Skills The presentation of information in this chapter assumes that you can already perform the tasks listed below. You can test your readiness to proceed by answering the Review Questions at the end of the chapter.

Chapter 11 Review Gases Mixed Answers  
CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: a. temperature increases, volume stays the same b. volume increases, temperature stays the same

Chapter 11 Review Molecular Composition Of Gases Section 2 ...  
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CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 224 L (b) divided by the mass of 1 mol (d) divided by 224 L 2 For the expression  $V = nRT/P$

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REVIEW Gases  
Chapter 11 Review Molecular Composition CHAPTER 11 REVIEW Molecular Composition of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in ...

Chapter 11 - Gases  
Chapter 11 Review Gases Answer Chapter 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L. Chapter 11 Review Gases Answer Key

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CHAPTER 11 REVIEW . Gases . SHORT ANSWER . Answer the followin9 questions in the space provided. 1. c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 . L. (b) divided by the mass of 1 mol. (d) divided by 22.4 . L. 2. c For the expression  $V = nRT/P$

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Chapter 11 Review Gases Section 4 Answers CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 224 L (b) divided by the mass of 1 ... REVIEW Gases - Weebly CHAPTER 11 REVIEW Gases SHORT ...

Chapter 11 Review Gases Mixed Answers  
CHAPTER . 11 . REVIEW . Gases . SHORT ANSWER . Answer the following questions in the space provided. 1. Consider the following data table: Approximate pressure (kPa) Altitude above sea level (km) 100 . o (sea level) 50 5.5 (peak of Mt. Kilimanjaro) 25 11 Get cruising altitude) < 0.1 22 (ozone layer) ~.

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Chapter 11 181 Chapter 11 - Gases Review Skills 11.1 Gases and Their Properties Ideal Gases ... Work all of the selected problems at the end of the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it.

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452 Chapter 11 Gases Discovering the Relationships Between Properties If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume, temperature, or number of gas particles) inside and outside the balloon.

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