

## Chapter 11 Chemical Reactions Answers Pearson

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Section 11.3 □ Reactions in Aqueous Solution. A complete ionic equation is an equation that shows dissolved ionic compounds as dissociated free ions. A spectator ion is an ion that is not directly involved in a reaction. Compounds that are aqueous will break into ions, and compounds that are solid will remain bonded.

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oxidation□reduction reactions A chemical reaction that exhibits a change in the oxidation states of one or more elements in the reactants that has the general form oxidant + reductant □ reduced oxidant + oxidized reductant. The general forms of these five kinds of reactions are summarized in Table 11.6.1, along with examples of each.

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Chapter 11 Chemical Reactions Answers

Chemistry chapter 11 Chemical reactions answer key. coefficient. a whole number that appears before a formula in an equation. spectator ion. a particle not directly involved in a chemical reaction. combustion reaction. a reaction in which oxygen reacts with another substance, often producing light or heat. reactant.

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Chapter 11: Chemical Reactions

Write correct formulas for the products in these double replacement reactions.  $\text{H}_3\text{PO}_4 + 2\text{K}_2\text{CO}_3 + \text{BaCl}_2 \rightarrow \text{Ba}_3(\text{PO}_4)_2 + 6\text{KCl}$   
 $\text{AgNO}_3 + \text{KCl} \rightarrow \text{AgCl} + \text{KNO}_3$   
 $\text{H}_2\text{SO}_4 + \text{Co} \rightarrow \text{CoSO}_4 + \text{H}_2$   
 $\text{BCl}_3 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + 3\text{HCl}$   
 $\text{CaS} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2\text{S}$   
 $\text{Ag}_2\text{S} + \text{H}_2\text{O} \rightarrow \text{Ag}_2\text{O} + \text{H}_2\text{S}$

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