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pentahalides more covalent than trihalides? Ans: The group 15 elements have 5 e-1 s in their valence shell. It is difficult to lose 3e-1 s to form E 3+ and even more difficult to lose 5e-1 s to form E 5+. Thus, they have very little tendency to form ionic compounds.

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The p-Block Elements Class 12 Notes Chemistry Chapter 7 I Group 15 Elements: The Nitrogen Family 1. In group 15 of the Periodic Table, the elements, nitrogen (7N), phosphorus (15P), arsenic (33As), antimony (51Sb) and bismuth (83Bi) are present. 2. (a) The elements of this group can exhibit various oxidation states ranging between -3 to+ [...]

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