

Bohr Model Practice Answers

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Use the following Bohr model to answer questions 1 to 6. Match the Term on the left with the corresponding Number on the right. Each Number may be used more than once. Refer to the diagram. Practice Draw Bohr models of the following ions. Include: symbol, protons, neutrons,

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Atomic Structure: Chapter Problems Bohr Model Class Work 1. Describe the nuclear model of the atom. 2. Explain the problems with the nuclear model of the atom. 3. According to Niels Bohr, which wavelengths emitted from a hydrogen atom, a student finds that the value of "n" in the Balmer Series of hydrogen is 4.5.

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I hand out the Bohr Model Practice guidelines and show the students how to use them. We complete the worksheet together. I also model the variation in which the number of electrons rather than the separate energy levels.

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Bohr Model Worksheet Directions Draw the Bohr Models showing all the electrons in each energy level. 1. Magnesium compounds are used in the production of uranium for nuclear reactors. Draw the Bohr model for magnesium. Niels Bohr 2. Sodium is found in salts that can be used to seed clouds to increase rainfall. Draw the Bohr model for sodium.

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Bohr Model Worksheet Use the description sheet and the periodic table to help you complete the following Bohr models. 1. How many electrons can each shell hold? a. 1st = _____ b. 2nd = _____

Atomic # Atomic Mass Protons Neutrons Electrons Bohr Model Carbon 6 12 6 6 6 Hydrogen 1 1 Lithium 3 3 3

Mrs. Cooks

The Bohr diagram is associated with the Bohr model, which depicts a small, positively charged nucleus surrounded by electrons that travel in circular orbits around it, similar to the structure of the

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Bohr Model Practice Problems Name: _____ Period: _____ In 1943, Niels Bohr described the atom as a planetary system with electrons orbiting around the nucleus. Bohr's model is incomplete, but it provides a better understanding of why some atoms are very reactive and others are less so.

Bohr Model Practice Worksheets - Lesson Worksheets

Bohr Model Practice For each element, write the total number of electrons on the line. Then color the correct number of electrons for each orbit. Remember, fill the orbit closest to the nucleus first. Each orbit can hold a certain number of electrons. Check the Periodic Table to find out how many electrons each element actually has.

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Check your understanding of Bohr Models and Atomic Spectra with this interactive quiz and printable worksheet. These practice assets will help you...

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Name: _____ KEY _____ Date: _____ Period: _____ Page 1 of 4 © 2004 High School Technology Initiative (HSTI) Educational Materials: The ATOM: Structure

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