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chapter 4: the structure of the atom Although the ancient Greeks had many creative ideas about matter, our current understanding is based upon John Dalton's atomic theory, published in 1803.

Essentially, matter is made up of atoms, which are indivisible and indestructible.

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### Class 7- Chemistry- Chapter 4- Atomic Structure- Part (3) Periodic Table

Atoms—and the protons, neutrons, and electrons that compose them—are extremely small. For example, a carbon atom weighs less than  $2 \times 10^{-23}$  g, and an electron has a charge of less than  $2 \times 10^{-19}$  C (coulomb). When describing the properties of tiny objects such as atoms, we use appropriately small units of measure, such as the atomic mass unit (amu) and the fundamental unit of ...

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Chapter 4 Atomic Structure 33 SECTION 4.1 DEFINING THE



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ATOM (pages 101 – 103) This section describes early atomic theories of matter and provides ways to understand the tiny size of individual atoms. Early Models of the Atom (pages 101 – 102) 1. Democritus, who lived in Greece during the fourth century B.C.,

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Chapter 4: Atoms and Elements Chapter Summary Chapter 4 starts with an overview of atomic theory. From the ancient philosophers to Dalton and on to Rutherford ' s experiments, the evolution of the atomic structure is concisely presented and connected to the ultimate “ cheat sheet ” in chemistry, the periodic table of elements.

STRUCTURE OF ATOM/CHAPTER 4/CLASS 9

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