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AP Chemistry Chapter 11. Intermolecular Forces, Liquids, and Solids - 10 - Vapor Pressure and Boiling Point • Liquids boil when the external pressure at the liquid surface equals the vapor pressure. • The normal boiling point is the boiling point at 760 mm Hg (1 atm).

AP Chemistry Test (Chapter 11)
As we saw in Chapter 4, the solvent is normally the component present in greatest amount. Other components are called solutes. Because liquid solutions are the most common, we will focus our attention on them in this chapter. Our primary goal is to examine the physical properties of solutions, comparing them with the properties of their components.

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PDF AP Chemistry Chapter 15 Answers - Zumdahl 15. AP Chemistry Chapter 15 Answers - Zumdahl 15.51 HA + OH-? A+ H 2O: Added OH-from the strong base converts the weak acid, HA, into its conjugate base, A- Initially, before any OH-is added (point d), HA is the dominant species present.

Chapter 11: Thermochemistry-Heat and Chemical Change
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AP Chemistry - Solutions, Chapter 11/4b FAQ. Q: What are hydrophobic and hydrophilic? A: Hydrophobic is a "fear of water". Nonpolar substances are typically hydrophobic. They are not attracted to water. Hydrophilic on the other hand are "water lovers". These compounds are attracted to water, and therefore are commonly ionic or polar.

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At a given temperature the vapor pressure of pure liquid diethyl ether and isopropyl alcohol are 730 torr and 310 torr, respectively. A solution prepared by mixing isopropyl alcohol and diethyl ether obeys Raoult's law.

Baker, Mrs. (Science) / AP Chemistry
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A.P. Chemistry Practice Test: Ch. 11, Solutions Name_____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Formation of solutions where the process is endothermic can be spontaneous provided that _____. A)the solvent is a gas and the solute is a solid

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