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lost by the  
other substance.  
Suppose you  
place 125 g of  
aluminum in a  
calorimeter with  
1,000 g of  
water. The water  
changes  
temperature by 2  
°C and the



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**Calorimetry**  
**DESCRIPTION.**

*Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature*

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values to see  
effects on the  
system. One or  
any combination  
of the  
substances can  
be mixed with  
water. A dynamic  
graph  
(temperature vs.

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place 125 g of  
aluminum in a  
calorimeter with**

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Example Question

#1 :

Calorimetry,  
Specific Heat,  
And  
Calculations. 1.  
The density of  
water allows us  
to say that 300  
milliliters of  
water is the  
same thing as

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300 grams of  
water. 2. Since  
the heat from  
the iron is  
being  
transferred to  
the water, we  
can say that the  
heat transfer is  
equal between  
both compounds.

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use the  
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Gizmo™** to  
determine the  
specific heat  
capacities of  
various  
substances. On  
the **SIMULATION  
pane, select**

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*Copper . Use the slider to set its Mass to 200 g. Set the Water mass to 200 g. Check that the Water temp is set to 30.0 °C and the copper's Temp is 90 °C.*

*Answer Key  
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be used to find  
relative  
specific heat  
values when  
different  
substances are  
mixed with  
water. Modify  
initial mass and  
...*

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science of  
measuring heat  
flow that occurs  
during a  
physical or  
chemical change.  
Heat: Form of  
energy that  
moves from one  
object to  
another because  
of a difference**

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in tempertaure.  
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When a hot  
object is placed  
in contact with  
a cold object  
heat flows from  
the hot object  
to the cold  
object until  
they reach?

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**Exploration:**

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**Vocabulary:**

**calorie,**

**calorimeter,**

**joule, specific**

**heat capacity**

**Prior Knowledge**

**Questions (Do**

**these BEFORE**

**using the**

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Latin word calor  
means "heat,"

and meter comes

from the Greek

word meaning "to

measure." What

do you think a

calorimeter

does? \_\_\_\_\_

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