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1 CHAPTER ONE

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Questions 16. a. Law
versus theory: A law is a
concise statement or

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equation that summarizes observed behavior. A theory is a set of hypotheses that gives an overall explanation of some phenomenon. A law summarizes what happens; a theory (or model) attempts to explain why it happens.

b.

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Answer. (a) Corrosion

It is a chemical process of decay of metal when they are exposed to moisture.

(b) Rusting
Corrosion of iron is called rusting. It is a slow oxidation reaction which takes place in the presence of moisture.

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Chapter 9 141 Section

9.1 Reactions and

Equations pages

282–288 Practice

Problems pages

284–287 Write skeleton

equations for the

following word

equations. 1. Hydrogen

and bromine gases react

to yield hydrogen

bromide.

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Chemical Equations and
Reactions CHAPTER 8
REVIEW Chemical
Equations and Reactions
Teacher Notes and
Answers Chapter 8
SECTION 1 SHORT
ANSWER 1. a. d b. a c.
b d. f e. e f. c 2. 8,4,9 3.
a. 12 atoms b. 16 atoms

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c. 51 atoms d. 3 1024
atoms

Chemical Reactions

And Equations –

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Reactions and Equations
page 6 are given below.

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or go for ?????? ??????
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**REVIEW Chemical
Equations and
Reactions**

Chapter 1 Chemical
Foundations. Section
1.1 Chemistry: An
Overview A main
challenge of chemistry
is to understand the
connection between the

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macroscopic world that we experience and the microscopic world of atoms and molecules.

You must learn to think on the atomic level. 2.

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Based on the molecular diagram, classify each change as physical or chemical. (Figure 1)

(Figure 2) (Figure 3) ...

Use your calculator to determine the answer to the following calculation: 1.0×10^{154}

...

6 Chemical Bonding

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Review 1. a. Law versus theory: A law is a concise statement or equation that summarizes

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SECTION 1 SHORT

ANSWER Answer the

following questions in

the space provided. 1. a

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A chemical bond
between atoms results
from the attraction

between the valence
electrons and of
different atoms. (a)

nuclei (c) isotopes (b)
inner electrons (d)

Lewis structures 2. b A
covalent bond consists
of (a) a shared electron.

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