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*amount of gas. $PV/T = k$. or . $P_1 V_1 /T_1 = P_2 V_2 /T_2$.
2. Charles' Law. Episode 903. Ex: A sample of gas has a volume of 201 L when its temperature is 293 K and its pressure is 224 mm Hg. What volume will the gas occupy at STP? Given:*

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Complete each of the following showing all work and circling your final answer on all problems. 1. To change a temperature expressed in degrees Celsius to a temperature on the Kelvin scale, what must be done to the Celsius temperature? Why must we use the Kelvin scale in gas law problems? 2. The volume of a sample of gas is 2.00 L when the ...

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