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REVIEW and REINFORCEMENT Section 6-3

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Reinforcement & Vocabulary Review Worksheets

3-1 Review and Reinforcement Early Models of the Atom On the line at the left, write the letter of the chemist who proposed each of the ideas listed below. Each letter may be used once, more than once, or not at all. ... 17. 12 protons and 10 electrons 18. 8 protons and 10 electrons

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CAEn0 .1 //omEL0ÖRK S 8-1 Review and Reinforcement, pages 12-13 8-2 Review and Reinforcement, pages 19-20 polar I polar 1. 2. 3. 5. 6. 8.

10-2 Review and Reinforcements - Front Page

17. The volume of gas particles themselves is assumed to be _____. 18. Gas molecules are said to be in _____. 19. The collisions between gas particles are _____. 20. The temperature of a gas is a measure of the average _____ of the gas ... 12-1 Review and Reinforcement ...

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'16-1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of a liquid placed in. a container, equilibrium is established when the Equid vaporizes at the same rate that vapor condenses. 7, NO, in reversible reactions at equilibrium,

12-1 Review and Reinforcement - North Penn School District

11-1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter coefficients subscripts mass

Chemistry: Chapter 15 Concentration and Nature of Solutions

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10-2 Review and Reinforcement (continued) 9. $0.0621 \text{ mol SS} = (Q \text{ } \$ \text{ gK2S } 10. 65.8 \text{gMg} = 1. \text{ atoms Mg}$ Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO₂ gas.

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11-I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid? The ... 17. Zinc reacts with H₂SO₄ to yield ZnSO₄ and hydrogen gas. How manv moles of hydrogen will

9 1 REVIEW AND REINFORCEMENT THE NATURE OF CHEMICAL ...

2 Answer.Key 2-1 Explore, page 5 ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed ... 17. distillation 18. A magnet could be used to separate them because iron is attracted to a magnet but

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REVIEW and REINFORCEMENT Chemical Families Section 6-3.... KEY CONCEPTS. - ~ Elements within the same family of the periodic table have simi(ar properties because they have the same number of valence electrons. M Vocabulary Skills: Applying Definitions On the periodic table below, label each of the following: acnoid series alkaline earth ...

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